

DAVITAYA, I.P., professor; KHOCHOLAVA, K.M.

Intracerebral pneumocephalus. Vop.neirokhir. 20 no.2:55-57 Mr-tp '56.
(MLRA 9:?)

1. Iz fakul'tetskoy khirurgicheskoy kliniki lechebnogo fakul'teta
Tbilisskogo meditsinskogo instituta
(BRAIN, dis.)

pneumocephalus caused by gunshot wound of brain)
(WOUNDS AND INJURIES
gunshot wound of brain causing pneumocephalus)

NOSACH, Ye.G.; KHOCHOLAVA, V.N., inzh.

Phase registration of the disconnection of short-circuits by
switches. Energetik 11 no.1:17-18 Ja '63. (MIRA 16:1)
(Electric power distribution). (Electric protection)

KHODA, V. (Odessa)

Coating the chassis with lacquer. Radio no. 6:56 Je '57. (MLRA 10:7)
(Radio--Apparatus and supplies)

USSR/Pharmacology. Toxicology. Tranquillizers

V

Abs Jour : Ref Zhur - Biol., No II, 1958, No 51905

Author : Galenko V. Ye., Khodabak A.I.

Inst

Title : Experimental Application of Serpasil in Psychiatric Diseases

Orig Pub : V. sb. Vopr. psichiatrii, Vyp 2, M., 1957, 36-58

Abstract : No abstract

Card : 1/1

KHODAK, A. N.

N/5

615.918

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APPROVED FOR RELEASE: 09/17/2001 CIA RDP86Q00512R000722120007-0
Nomenklatura opisanii po voprosam poverkhnostnoy zakalke (Classed List of Equipment Subject to Flaming Case Hardening, by) M. V. Naboka, P. I. Rodzevich, A. H. Khodak. Khar'kov, metallurgizdat, 1952.

71 p.

At head of title: Russia. Nauchnoissledovatel'skoye byuro organizatsiy proizvodstva chernoy metallurgii.

KHODAK, A.N.

RAYKO, V.V.nauchnyy sotrudnik; VOLKOV, Ya.R.nauchnyy sotrudnik; LEVITSKIY,
D.A.nauchnyy sotrudnik; KHODAK, A.N.nauchnyy sotrudnik; RATNER, Yu.Z.
inzhener; VORODIMOV, N.I.inzhener; GRISHAYEV, N.N.inzhener;
SHULYATSKIY, D.I.,inzhener, redaktor; ANDREYEV, S.A.,tekhnicheskiy
redaktor

[Rules for the technical operation of cranes] Pravila tekhnicheskoi
ekspluatatsii pod" emnykh kranov. Khar'kov, Gos. nauchno-tekhn. issd-
vo lit-ry po chernoi i tsvetnoi metallurgii, 1957. 167 p.
(MLRA 10:5)

1. Russia (1923 U.S.S.R.) Ministerstvo chernoy metallurgii.
2. Vsesoiuznyy nauchno-issledovatel'skiy institut organizatsii
chernoy metallurgii. (for Rayko, Volkov, Levitskiy, Khodak)
3. Otdel glavnogo mekhanika Ministerstva chernoy metallurgii. (for
Shulyatskiy) 4. Zavod "Azovstal'" (for Ratner) 5. Zavod "Zaporozhstal'"
(for Vorodimov, Grishayev)
(Cranes, derricks, etc.)

RAYKO, V.V., nauchnyy sotrudnik; NIKBERG, I.M., nauchnyy sotrudnik;
KHODAK, A.N., nauchnyy sotrudnik; NEVEDUSHCHIY, A.I., nauchnyy
sotrudnik; VOLKOV, Ya.R., nauchnyy sotrudnik; FEYCHEV, G.P., o'tv.
red.; IPATOV, P.P., red.; SHULYATSKIY, D.M., red.; BURKSER, L.D.,
red.; BALASEVICH, Yu.Yu., red.; SVETCHENKO, V.N., red.; KRYLOVSKIY,
A.P., red.; SINYAVSKAYA, Ye.K., red.izd-va; ANDREYEV, S.P., tekhn.red.

[Regulations for operating the mechanical equipment of rolling mills]
Pravila tekhnicheskoi ekspluatatsii mekhanicheskogo oborudovaniia
prokatnykh tsekhov. Khar'kov, Gos.nauchno-tekhn.izd-vo lit-ry po
chernoi i tsvetnoi metallurgii, 1959. 247 p. (MIRA 12:9)

1. Moscow. Vsesoyuznyy nauchno-issledovatel'skiy institut organi-
zatsii proizvodstva i truda chernoy metallurgii. 2. Vsesoyuznyy
nauchno-issledovatel'skiy institut organizatsii proizvodstva i truda
chernoy metallurgii (VNIIOCHEMET) (for Rayko, Nikberg, Khodak, Neve-
dushchiy, Volkov). 3. Otdel glavnogo mekhanika byvshego Ministerstva
chernoy metallurgii SSSR (for Ipatov, Shulyatskiy). 4. Zavod imeni
Dzerzhinskogo (for Burksner, Balasevich). 5. Zavod imeni Kirova (for
Svetchenko). 6. Zavod imeni Voroshilova (for Krylovskiy).

(Rolling mills--Equipment and supplies)

KHODAK, A.S., (Bryansk).

Determination of small concentrations of ethyl alcohol. Apt.delo 2 no.2:
69-71 Mr-Ap '53.
(MLRA 6:5)
(Alcohol, Denatured)

(CA 47 no.16:7949 '53)

Khodak, A. S.

Rapid method of detection of methyl alcohol in different materials. A. S. Khodak. *Aptekarskoe Delo* 4, No. 6, 30-31 (1959).—An app. is described by which the MeOH vapors of the boiling material are absorbed in either 2% $K_2Cr_2O_7$ in 25% H_2SO_4 , or 3% $KMnO_4$ in 5% H_2SO_4 , to be oxidized to $HCHO$. The latter is carried over to a soln. of a few grains of either morphine, cocaine, or dionine in 2-3 cc. of H_2SO_4 . In the presence of H_2O_2 a red-violet coloration changing to violet and forming a ring between the two liquids takes place. The limit of sensitivity is 0.2 cc./100 cc. In case of alc. or alc. fluids 2-10 cc. dil. to 80-100 cc. and in case of urine 100-200 cc. previously acidified with citric or tartric acid are taken for analysis. In chem.-legal cases 100-200 g. of cadaver material crushed and worked into a paste with either acidified urine or dilut. water are used.

A. S. Minkin

CH

KHODAK, A.S.

**Quantitative determination of small concentrations of ethyl alcohol
in biological materials and pharmaceutical preparations. Apt.delo 6
no.1:42-45 Ja-F '57. (MLRA 10:3)**

**1. Iz byuro sudebno-meditsinskoy ekspertizy Bryanskoy oblasti
(nachal'nik S.P.Zubkov)
(ETHYL ALCOHOL)**

BRODSKIY, I.I., inzh.; GNILENKO, B.A.; KRYUKOV, G.Ya.; MARSHAK, V.I.;
KHODAK, I.Z.

Modernization of a continuous pipe-rolling mill. Mekh.i avtom.
proizv. 14 no.1:24-26 Ja '60. (MIRA 13:5)
(Pipe mills)

KHODAK, K. M.

KHODAK, K. M.: "The functional state of the higher portions of the central nervous system in various phases and in various clinical forms of rheumatism in children." Khar'kov Medical Inst. Khar'kov, 1956.
(Dissertation for the Degree of Candidate in Medical Sciences).

SO: Knizhnaya letopis', No 23, 1956.

CA

4

Mechanism of the anodic overvoltage in the electrolysis of cryolite-alumina melts. R. I. Remppel and L. P. Khodat Dolody Akad. Nauk S.S.R.A. 78, 853-854 (1950).—The theories ascribing the high potential E of C anodes at high c.d., observed in the electrolysis of salts of $Al(O)$ in fused $NaAlF_6$, to accumulation of O_2 to a high partial pressure (e.g. Pearson and Waddington, J.A. 48, 4074a) and subsequent oxidation of the C anode by O_2 at a secondary, purely chem. (not electrochem.) process, are contradicted by oscillographic observations of the change of the anode potential E after interruption of the polarizing current. These detns. show potentials of well over 1 v. to persist for several sec.; the quantity of electricity discharged in a fall of E from 1.6 to 1.84 v. shows the amt. of electrochemically active O_2 to be thousands of times the amt. calcd. for the gas phase.

With increasing c.d., E of a fresh unpolarized C anode increases exponentially, whereas it remains practically const. for a long time when the c.d. is lowered from 0.1 to 0.3 amp./sq. cm. Thus, E cannot be ded. by the partial pressure of gaseous O_2 . Inasmuch as C does interact chemically with O_2 ($\sim 930^\circ$), the C-O₂ electrode cannot be considered as a gas electrode analogous to the C-Cl₂ electrode where there is no chem. interaction between C and Cl₂. Under a const. c.d., oscillograms of E often show a max. In the light of the known facts on the combustion of C, the most probable effect of the discharge of oxygenated anions C is the formation of intermediate C-O compds. at the surface of C. Bond regrouping processes in this surface compd. are relatively slow, and the final stage is decompr. of the intermediate products and desorption of gaseous C oxide where O is most strongly bound. Accordingly, at low c.d., the decompr. potential, 0.08 v., is very close to the reversible potential of $Al(O) + 3 C \rightleftharpoons 2 Al + 3 CO$. With the discharge extending, at higher c.d., to a point of lower heat potentials, and the more loosely held O ions will pass into the oxides formed at high c.d. and high E which accounts for the practical absence of a change of E on lowering the c.d. The slow process is the decompr. of the intermediate surface C-O compds., as is known from combustion of C. It is not the stage of discharge of anions and of formation of the adsorption complexes. Then the

1951

KHODAK, L.P.

✓ Mechanism of the origin of overvoltage on a carbon anode
in cryolite aluminum oxide melts. S. I. Rempel and L. P.
Khodak. Zhur. Priklad. Khim., 26, 931-40 (1953); cf.
C.A. 43, 6519c. —A detailed discussion supported by many
references to the work of others and to exptl. data obtained
by the authors in this and previous publications are pre-
sented to refute the theory of Drossbach (C.A. 30, 4095)
and his "followers" that a C anode in an Al_2O_3 bath is a
reversible O electrode. The specific capacity of the double
layer on a "fresh" C electrode is 15-20 microfarads/sq. cm.
During and after electrolysis it is appreciably greater. It is
concluded that neither concd. polarization nor retardation of
the discharge reaction are primary factors of appreciable
overvoltage. The discharge of O-contg. ions on a C elec-
trode results in the formation of "chemisorbable C-O com-
plexes" and not free O. The low rate of decompn. of these
complexes and the heterogeneity (energy) of the C surface
cause the increase of the anodic potential with the increase
in c.d., i.e. the primary cause of overvoltage on a C anode is
the low rate of formation of gaseous O oxides from the com-
plexes and their subsequent desorption from the surface of
the anode.

I. Bencowitz

SHODAK, I.A.P.

USSR,

*/ The Casse of Alumina smelting and its connection with
Current Electrolytic S. I. Kurnik and I. P. Gerasimov
Prilozh. Akad. 1954, 87, 3. The Telegraphic Trans.
R. and R. reply to questions
W.M. Johnson, Prof. Engg., Dept. of Metallurgy,
Massachusetts Institute of Technology, Cambridge,
Mass., U.S.A.*

Electrometallurgy of Aluminum, Moscow 1953

*Inst. Chem. + Metallurgy
Ural. Offt. AS USSR*

KHODAK, L. P.

USSR/Chemistry - Physical chemistry

Card 1/1 : Pub. 22 - 27/48

Authors : Rempel', S. I.; Anisheva, N. A.; and Khodak, L. P.

Title : Comparison gas-electrode for measurements of cryolite-alumina fusions

Periodical : Dok. AN SSSR 97/5, 859-862, August 11, 1954

Abstract : The characteristics of various gas comparison-electrodes, used for the measurement of cryolite-alumina fusions, are analyzed. The oxygen-carbon electrode is considered to be the most stable comparison electrode and because of its high accuracy is best recommended for measurements of cryolite-alumina fusions. Means of securing composition constancy of the gaseous mixture surrounding the comparison electrode and to prevent anode gases from falling into the gas mixture, are described. Five USSR references (1944-1953). Graph; drawing.

Institution : Acad. of Sc. USSR, Ural Branch, Institute of Chemistry and Metallurgy

Presented by : Academician A. N. Frumkin, April 3, 1954

KHODAK, L. P.

Measurement of the anodic percentage in the electrolyte of
graphite anode during the discharge of the battery

With increasing c.d., R of a fresh carbonated C anode increases exponentially where
for a long time there is no change in current density. Then the current density increases
with the increase of pressure of gas. This is due to the fact that the diffusion coefficient of
carbon dioxide in the electrolyte is very small. When there is no diffusion of gas, the
current density is constant.

After the completion of the first cycle of discharge and charge, the capacity of the battery
is reduced by about 10% due to the formation of a thin layer of electrically conductive
oxides on the surface of the electrodes.

During the second cycle of discharge and charge, the capacity of the battery is reduced by
about 10% due to the formation of a thin layer of electrically conductive
oxides on the surface of the electrodes.

1951

(Handwritten text)
SUSHKOV, Akim Ivanovich; TROITSKIY, Ivan Alekseyevich; NYDENZON, Moissey
Aronovich; KHODAK, L.P., kand.tekhn.nauk, red.; IVANOV, A.I., inzh.
red.; REMPEL', S.I., prof. doktor tekhn.nauk, red.; LUCHKO, Yu.V.,
red.izd-va; ZEF, Ye.M., tekhn.red.

[Metallurgy of light metals] Metallurgiia legkikh metallov.
Sverdlovsk, Gos.nauchno-tekhn.izd-vo lit-ry po chernoi i tsvetnoi
metallurgii, Sverdlovskoe otd-nie, 1957. 510 p. (MIRA 11:2)
(Light metals--Metallurgy)

SOV/137-58-12-24284

Translation from: Referativnyy zhurnal. Metallurgiya, 1958, Nr 12, p 51 (USSR)

AUTHOR: Khodak, L. P.

TITLE: The Principle Underlying the Plotting of the Energy Balance of a Cell
(K voprosu o printsipe postroyeniya energeticheskogo balansa elek-
trolizera)

PERIODICAL: Tr. In-ta metallurgii. Ural'skiy fil. AN SSSR, 1957, Nr 1, pp
139-143

ABSTRACT: A correlation of the processes proceeding in pyrometallurgical installations and electrolysis baths (EB) leads to the conclusion that it is impossible to apply the principle employed in compiling the heat balance of metallurgical furnaces (addition of exothermic processes within the installation to the balance) to that of EB energy balances. In the energy balance of an EB the major item of input is the energy of the electric current, while the major output items are the energy requirements for increasing the heat content of the flow of material passing through the EB (including that required for initial heating of the materials) and the energy required for the compensation of the heat loss into the ambient medium.

Ye. Z.

Card 1/1

SOV/137-59-1-460

Translation from: Referativnyy zhurnal. Metallurgiya, 1959, Nr 1, p 58 (USSR)

AUTHORS: Khodak, L. P., Rempel', S. I., Kuznetsov, S. I.

TITLE: On the Effect of Periodic Charging of Raw Materials on the Energy Balance in an Electrolytic Bath (O vliyanii periodicheskikh zagruzok syr'ya na energeticheskiy rezhim elektroliznoy vanny)

PERIODICAL: Tr. In-ta metallurgii. Ural'skiy fil. AN SSSR, 1957, Nr 1, pp 144-148

ABSTRACT: Utilizing the case of an Al bath as an example, the authors analyze the significance of the heat of dissolution of raw materials introduced into the bath in batches and the significance of an increase in electrical potential occurring in the bath during periods between charging on energy-balance equations for an electrolytic bath. In computing variations in heat content of a system undergoing an electrochemical reaction, e.g.: $\text{Al}_2\text{O}_3 + 1 - \frac{1}{2}\text{C} \rightarrow 2\text{Al} + 1 - \frac{1}{2}\text{CO}_2$, the right-hand side of the equation must show the Al_2O_3 in the form in which it is introduced into the bath rather than in the form of Al_2O_3 already dissolved in the electrolyte. In computing the energy balance, the alumina (A) must appear in the same form as that

Card 1/2

SOV/137-59-1-460

On the Effect of Periodic Charging of Raw Materials on the Energy Balance (cont.)

in which it appears in the material balance. An increase in the electrical potential of the bath, which occurs as the concentration of A in the electrolyte is reduced, results in the consumption of an additional quantity of electrical energy Q_g . If the increase in potential is brought about merely by a change in the concentration of A, then $Q_g = \Delta Z_p$, where ΔZ_p is the change in the isobaric-isothermal potential of the system. If other causes also are active, the magnitude of the ΔZ_p will be fully compensated by the electrical energy supplied externally. The magnitude of the Q_g will be the sum of two items: $Q_g = \Delta Z_p + Q'$, where Q' represents the additional quantity of electrical energy consumed in the bath as a result of an increase in the potential due to any causes aside from a change in the concentration of the A. Being a separate item on the input side of the energy-balance equation, the heat of dissolution of the raw material in the electrolyte must not be taken into consideration. A method permitting computation of the mean increase of potential in electrolyzers operating with raw material introduced periodically is given together with the computation of the mean value of the potential increase in an Al bath (this value being 0.105 v in said instance). In computing variations in the heat content of a system undergoing an electrochemical reaction the output side of the balance equation must be based on thermodynamic data for solid rather than dissolved A.

L.S.

Card 2/2

KHODAK, L.P.; REMPEL', S.I.; KUZNETSOV, S.I.

Energy balance of aluminum bath. Trudy Ural.politekh.inst.
no.58:88-96 '57. (MIRA 11:4)
(Aluminum--Electrometallurgy)

KHODAK, L. P.

400-177

REPORTS: Soviet Light Metal Industry Series, Vol. 2) Moscow, 1958, 280 p. (Berlin: TGA, 1960 English version).

Editorial Board: M. A. Al'tshuler, Yu. P. Berezovskiy, V. I. Kostyuk, A. G. L. Danziger, S. D. Gavrilov, N. N. Krasil'shchikov, and Ye. I. Khodak (Bulg. Ed.) Committee of Technical Estimates; Mr. of Publishing House: V. A. Solntsev, Prof. M. I. Fedotov.

PURPOSE: This issue of the Soviet Siberian Metal Transactions is of interest to economists, engineers and metallogists, metallurgists, and metallurgists in the light metal industry.

CONTENTS: This collection of articles is a compilation of the reports presented at the third international conference on the Chemistry of Light Metals organized by the Laboratory of Light Metallurgy of the Siberian Division of the USSR Academy of Sciences in Novosibirsk in October 1956. It is the first report of scientific cooperation between the metallography of the Soviet Union, Germany, and the East German metallurgy, light metals industry, and mining industries. The reports indicate that large aluminum and titanium reserves exist in the Soviet Union. They also provide the chemical analysis of soil and aluminum ore. Individual articles also report on the following subjects:

Chemical reactions in the development of the light metals industry in Siberia, aluminum and magnesium extraction, smelting, magnetite, etc., aluminum production, etc.

TABLE III. METALS AND THEIR PROPERTIES

Khodak, L. P. New Data on the Properties of Light Metals

Solntsev, V. V. Application of the Method of X-ray Diffraction to the Analysis of Light Metals

Khodak, L. P. Some Problems of Light Metals

Khodak, L. P. and A. I. Kostyuk. Combined Treatment of Aluminum Oxide from the Urals and Donets Basins

Khodak, L. P. and A. I. Kostyuk. Combined Treatment of Aluminum Oxide from the Donets Basin

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SECTION I. PLATE I. DOCUMENTATION 307/238

Khodak, L. P. Photocopy-Sterling file

REPORTS: Soviet Light Metal Industry Series, Vol. 2) Moscow, 1958, 280 p. (Berlin: TGA, 1960 English version).

Editorial Board: M. A. Al'tshuler, Yu. P. Berezovskiy, V. I. Kostyuk, A. G. L. Danziger, S. D. Gavrilov, N. N. Krasil'shchikov, and Ye. I. Khodak (Bulg. Ed.) Committee of Technical Estimates; Mr. of Publishing House: V. A. Solntsev, Prof. M. I. Fedotov.

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KHODAK, L.P.; IVANOV, A.I.

Using blast furnace slags for the production of alumina. Trudy Inst.
met. UFAV SSSR no.2:253-256 '58. (MIR 12:4)
(Alumina) (Blast furnaces--By-products)

KHODAK, L.P.; IVANOV, A.I.

Complex processing of alumina iron ores and high-iron bauxites.
Trudy Vost.-Sib. fil. AN SSSR no.13:237-241 '58. (MIRA 12:12)

1.Ural'skiy filial AN SSSR.
(Iron ores)

PHASE I BOOK EXPLOITATION 507/2216

5(4)

KHODAK L. P.
ELIZAVETIN N. G.

Soveshchaniye po elektrokhimii. 4th. Moscow, 1956.
 Trudy . . . [laboratori] (Transactions of the Fourth Conference on Electrochemistry). Collection of Articles. Moscow, 1st to AN SSSR.
 1959. 868 p. Arran. zhurn. SSSR. Otdeleniye khimicheskikh nauk. Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk.

Editorial Board: A.M. Frankin (Resp. Ed.) Academician, O.A. Yesin, Professor; S.I. Zhdanov (Resp. Secretary); B.M. Kabanov, Professor; V. M. Kolotyrkin, Doctor of Chemical Sciences; V.V. Losov, Professor; Iakorsey, Professor; Z.A. Solov'yeva, V.V. Stender, Professor; and G. M. Florinovich; Ed. of Publishing House: M.G. Tegurov; Tech. Ed.: T.A. Prusakova.

PURPOSE: This book is intended for chemical and electrical engineers, physicists, metallurgists and researchers interested in various aspects of electrochemistry.

COVERAGE: The book contains 127 of the 138 reports presented at the Fourth Conference on Electrochemistry sponsored by the Department of Chemical Sciences and the Institute of Physical Chemistry of the Academy of Sciences USSR. The collection pertains to different branches of electrochemical kinetics, double layer theories and advanced processes in metal electrodeposition and industrial electrolysis. Abridged discussions are given at the end of each division. The majority of papers have not been published elsewhere. References are given at the end of most of the articles.

Ponosenko, A.S., T.M. Abramsova and I.I. Dankina (Institut prirodnykh materialov DSSR-Institute of Physical Chemistry AS URSR). Mechanism of the Corrosion of Iron, Magnesium, Zinc and Aluminum With the Aid of Heavy Oxygen Ions 299

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Mashkov, Yu. V. and A.A. Revayyan (Vestnizhny aluminivodnogo in-ta imeni V.I. Vernadskogo-Alumina-Magnesium Institute). Mechanism of Anion Dissociation in the Equilibrium of Molten Cryolite Clay 334

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Card 13/ 3a

Card 14/ 3a

KHODAK, L.P.; KUZNETSOV, S.I.; IVANOV, A.I.; SEHGBRENNIKOVA, O.V.;
MOLEVA, N.G.

Obtaining alumina from blast furnace slags rich in the compound.
Izv.Sib.otd.AN-SSSR no.2:19-28 '59. (MIRA 12:7)
(Alumina) (Slag)

MOLEVA, N.G.; IVANOV, A.I.; XHODAK, L.P.

Effect of the calcium oxide content on the structure and properties of
easily crumbling aluminum-calcium slags. Izv. Sib. otd. AN SSSR no.8:
58-61 '59. (MIRA 13:2)

1.Ural'skiy filial AN SSSR.
(Slag)

KHODAK, L.P.; VARLAMOVA, N.N.; KOZHEVNIKOV, G.N.

Extraction of alumina and alkali from sinters obtained in the
reduction smelting of red muds. Izv. Sib. otd. AN SSSR no.7:
64-70 '62

1. Ural'skiy filial AN SSSR, Sverdlovsk.

GRUZINOV, V.K., akademik; MIKHAYLOV, V.V., akademik; KHODAK, L.P., kand.
tekhn.nauk; MIKHAYLOV, S.V.; RAKHIMOV, A.R.; NIKITIN, G.M.

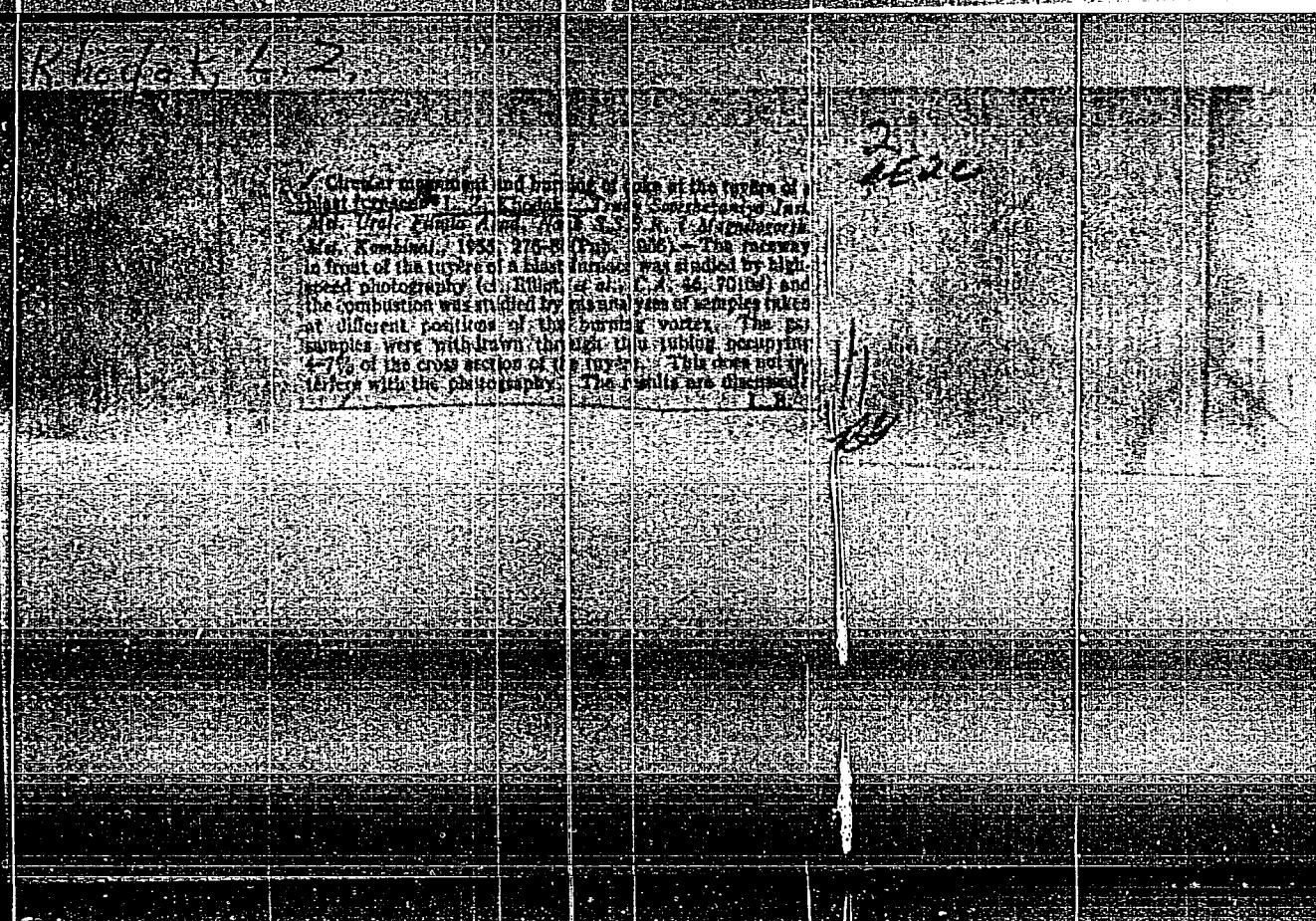
Utilization of the brown ores of the Lisakov deposit. Vest.
AN Kazakh. SSR 21 no.11:9-13 N '65.

(MIRA 18:12)

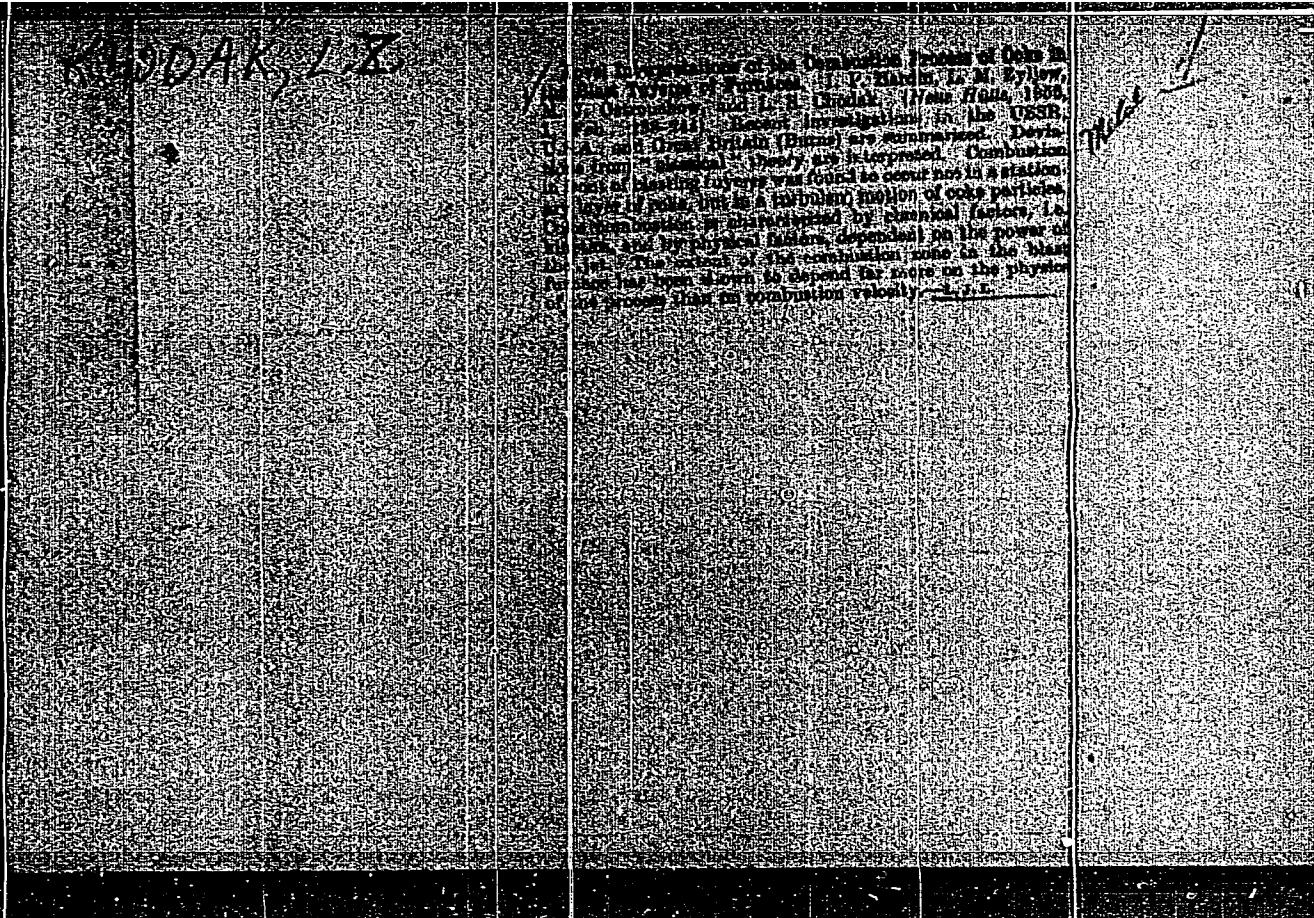
1. Akademiya nauk Kazakhskoy SSR (for Gruzinov, Mikhaylov, V.V.).

APPROVED FOR RELEASE: 09/17/2001

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M. D. A. I. T., L. Z.

1/15/62

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1817. COMBUSTION PROCESSES AT BLAST FURNACE TUVEYER. Ostroumoff, N. Ya.
and Khodak, L.Z. (Sov. Metal. Moscow), 1955, vol. 16, 867-874 abstr. in

Chem. Abstr., 1957, vol. 51, 2193). By means of small water-cooled pipes occupying 4-7% of the tuyere opening and involving packing in the tuyere stock, samples of gases in the combustion area were obtained. Combustion takes place in a cavity made by the blast in the charge and surrounded by a compact layer of coke 100-200 mm thick. Stereoscopic high speed photography shows that air entering this cavity from above is circulated by the blast in two loops. In the horizontal, the blast swirls around the tuyere and is carried off at the tuyere plane along the cavity wall returning in the vertical loop to the tuyere, which practically does not extend below the tuyere level. Distribution of air in the plane of the tuyere. Distribution of air in the plane of the tuyere. They do not mix with the air entering from the tuyere. They do not mix with the air entering from the tuyere.

For more information refer to the original document, measured with a platinum resistance thermometer.

The author is of the opinion that the factor of air distribution in the tuyere is only one of the factors.

PP 6/10

OSTROUKHOV, M.Ya., kandidat tekhnicheskikh nauk; KHODAK, L.Z.; LEBEDEV, V.V.

Cinematographic study of the process of coke burning. Priroda 45 no.7:
78-81 Jl '56. (MIRA 9:9)

1. Institut metallurgii imeni A.A.Baykeva Akademii nauk SSSR, Moskva
(for Khodak). 2. Laboratoriya nauchno-prikladnye fotografiy kinematografii
Akademii nauk SSSR, Moskva (for Lebedev).
(Coke) (Combustion) (Cinematography--Scientific applications)

KHODAK, L. Z.

137-1958-2-2393

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 2, p 27 (USSR)

AUTHORS: Bardin, I.P., Tsylev, L.M., Ostroukhov, M.Ya., Khodak, L.Z.

TITLE: On the Process of Coke Combustion at the Tuyeres of a Blast Furnace (O protsesse gorenija koksa u form domennoy pechi)

PERIODICAL: Tr. In-ta metallurgii AN SSSR, 1957, Nr 2, pp 3-8

ABSTRACT: In 1954-55, in different regions of the Soviet Union, a study was made on six blast furnaces having effective volumes of 330-1386 m³. Gas samples were taken along the axis and above and below the axis of a tuyere. The diagram depicting the change in gas composition in the combustion zone differed markedly from the "classical diagram." From the path of the isorhythmic lines for CO₂, CO, and O₂ in a vertical plane it was possible to establish the direction of the blast and the pattern of circulation of the coke particles. These experiments led to the conclusion that combustion of the coke does not occur in the bed layer but inside the blast. In addition, the focal combustion zone was found to be distributed along a spherical surface nearly at the boundary of the combustion zone. The length of the oxidation zone was determined basically by the kinetic energy of the blast and did not depend appreciably on other factors.

G.Ch.

Card 1/1

1. Coke--Combustion 2. Blast furnaces--Applications

RHOOK, L. Z.

卷四

25(6) - **STATE v. ROBERT HOLLOWAY**
- **Defendant's Appeal.** Lasting challenge!

SYNTHETIC POLYMERS: Some new is of interest to physicians is mortality, as well as to medical personnel of the orthopaedic industry.

SYNTHETIC POLYMERS: Both volume of the Study (synthesis) of the Doctor's methods. Daniel A. Johnson (Orthopaedic Institute Inc., A. D. Johnson) presents on synthesis of polyesters, individual acrylic and chloro, and polyvinylchloride. Some of the studies pertain to the effects of these substances, the viscosity and other characteristics of plastic surfaces during dissolution in solvents, resulting of metals due to corrosion, characteristics

La pertinencia de estos datos es evidente en tanto que permiten establecer la existencia de una relación entre el desarrollo de las industrias y la evolución de la población. La población es un factor fundamental en el desarrollo económico, ya que la mano de obra es necesaria para la producción de bienes y servicios. La población también tiene un efecto directo en la demanda interna, lo que implica que un aumento en la población puede impulsar el crecimiento económico.

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President, American Society of Natural History; Vice-Chairman, National Research Commission; and U.S. Senator.

Effect of Trace Components on Physical Properties of Blast Furnace Slag

and the Committee on the Constitution of the State of Oregon.

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APPROVED FOR RELEASE: 09/17/2001

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KHODAK, I. Z.

18(0) PLATE 1 BOOK EXPLORATION 857/1728
Andorlyns'kij SSSR. Institut metallicheskij

Sovremennye problemy metallichestva (Modern Problems in Metallurgy).
Moscow, Izd-vo Akademii Nauk SSSR, 1958. 640 p., 3,000 copies printed.

Book, Dr. A.M. Samarskiy, Corresponding Member, USSR Academy of
Sciences; Prof. V.S. Polubikhin, Moscow V.G. Kurchatov, and
A.M. Semenov, Prof. Tech. Ed. T.V. Polyakova.

INTROD.: This book is intended for scientific and technical per-
sonnel in the field of metallurgy.

CONTENTS: This is a collection of articles on certain aspects of
metallurgy. The book is dedicated to Academician
Ivan Pavlovich Brailov on the occasion of his 75th birthday. The
book is divided into seven parts. The first part consists of
two articles presenting a brief account of the biography and
professional activity of the Soviet metallurgist. It includes an
article by James Chapman, Nicholas Great, and John Elliott (U.K.),
describing their meeting with Brailov in Moscow and also his
visit to the United States. The second part consists of three
articles and deals with raw materials and fuels for the Soviet
metalurgical industry. The third part represents the major
portion of the book. It consists of 25 articles dealing with
the various aspects of the metallurgy of pig iron and steel.
The fourth part consists of five articles treating the metal-
lurgy of nonferrous metals. The fifth part consists of three
articles on the forming of metals. The sixth part consists of
eight articles discussing certain aspects of physical metal-
lurgy. The last part deals with general problems in the field
of metallurgy. References are given after each article. No
bibliography is given.

TABLE OF CONTENTS:

Modern Problems in Metallurgy 857/1729
Ostreshchikov, N.P. and I.Z. Khodak [Candidates of Technical
Sciences], Metalurgical Institute Izhevsk A.A. Barkov, AS USSR].
The Thermal Condition of the Earth in Connection With the
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Semenov, P.A. [Engineer, Gidromet, Glazernoye (State Institute for the
Design and Planning of Metallurgical Plants). Effectiveness
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Prof. F. A.Ye. [Professor, Doctor Economic Sciences, Committee
on the Study of Production Forces, AS USSR]. Effect of
the New System of Pig-Iron Production in Elektrostal'
Furnaces in Western Siberia and the Far East [separate] on the
Actions. Moscow.

Khodak, I.Z. [Doctor of Technical Sciences], O.I. Arhipov
[Candidate of Technical Sciences], and V.I. Korotchikov
[Candidate of Technical Sciences]. Use of Ore High-melting
Furnaces. Card 7/2

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KHODAK, L. Z., Candidate of Tech Sci (diss) -- "Processes occurring in the combustion zone of a blast furnace". Moscow, 1959. 26 pp (Acad Sci USSR, Inst of Metallurgy im A. A. Baykov), 150 copies (KL, No 21, 1959, 117)

TSYLEV, Leonid Mikhaylovich; OSTROUKHOV, Mark Yakovlevich; KHODAK, Leonid Zalmanovich; ZINGER, S.L., red.izd-va; ATTOPOVICH, M.I., tekhn.red.

[Process of coke combustion in blast furnaces] Protsess gorenija koksa v domennoi pechi. Moskva, Gos.izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1960. 98 p.

(MIRA 13:5)

(Blast furnaces--Combustion) (Coke)

BAZILEVICH, Sergey Vladimirovich; LAZAREV, Boris Leonidovich; STARIKOV,
Modest Andreyevich; GOLOSKOV, Boris Viktorovich; KULIKOV, I.S.,
kand.tekhn.nauk., retsenzент; KHODAK, L.Z., red.; CHAPAYKINA,
F.K., red.ind-va; NATLYUK, R.M., tekhn.red.

[Methods for experimental investigation of the blast-furnace
process] Metody eksperimental'nogo issledovaniia domennogo
protsessa. Sverdlovsk, Gos.suchno-tekhn.izd-vo lit-ry po
chernoi i tsvetnoi metallurgii. Sverdlovskoe otd-nie, 1960.
(MIRA 14:3)
254 p.
(Blast furnaces) (Cast iron--Metallurgy)

OSTROUKHOV, M.Ya. (Chelyabinsk), KHODAK, L.Z. (Moskva)

Analysis of the coke combustion process in blast furnaces by fur-
nace gas constitution diagrams. Izv. AN SSSR, Otd. tekhn. nauk. Met.
i topl. no.6:5-13 N - D '60. (MIRA 13:12)
(Blast furnaces--Combustion)

S/133/60/000/012/015/015
A054/A027

AUTHORS: Dashevskiy, Ya.I., and Khodak, L.Z.

TITLE: All-Soviet Meeting of Furnacemen and Agglomeration Workers

PERIODICAL: Stal', 1960, No. 12, pp.1156-1157

TEXT: The All-Soviet Meeting of Furnacemen and Agglomeration Workers convened by the State-Plan Committee of the Council of USSR Ministers, the State Scientific Research Committee of the Council of USSR Ministers and the Scientific-Technical Association of Iron and Steel Metallurgy was held in Magnitogorsk from 25 to 29 October 1960. All the Soviet metallurgical plants, scientific research institutes and planning bureaus were represented at the meeting. Presidential address was given by P.I. Korobov, Vice-President of the GNTK (GNTK) of the USSR Council of Ministers and dealt with the present state of the blast furnace and agglomeration industry in the Soviet Union. The main topics of the section "ore-dressing and agglomeration" were: intensification of agglomeration and improving the quality of the agglomerate; application of heated and oxygen-enriched air in the agglomeration of iron ores; cooling the aggregates; introducing the most up-to-date types of agglomerating equipment; automation of agglomeration; developing pelletisation. Besides general remarks made to stress the need of improving enriching and agglomeration and more

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A054/A027

All-Soviet Meeting of Furnacemen and Agglomeration Workers

efficient machinery for these processes, it was proposed that: the upper limit of particle size of the agglomerated ore be lowered to 6-8 mm, that of fuel and fluxes for agglomeration to 1 mm; a uniform quality of ores and concentrates, fuels and fluxes be produced; the amount of lime in the agglomeration charge be raised to 4-5%; the process of lime calcination be intercalated in the agglomeration technology; the weight system be introduced in the dosage of the charge and all plants be supplied with the appropriate equipment between 1961 and 1963; the mixing of agglomerate charge be improved through two-stage mixing, the mixing drums extended, etc; the amount of air pumped through the agglomeration charge be increased to $100^3/m^2$ of agglomerate; the agglomerizing equipment be fed with two charges having different fuel contents in the beds; hot air and combined heating be applied during agglomeration; special cooling equipment be constructed for the production of agglomeration fuel from weakly agglomerizing coal and peat; the agglomerate be air-cooled by forced draft, two agglomerizing machines with automatic control be installed; high-capacity agglomerating machines be designed and constructed, new technological processes be established, etc. With regard to blast furnaces the following problems were discussed: theoretical and experimental investigation of the combined blasting method;

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A054/A027

All-Soviet Meeting of Furnacemen and Agglomeration Workers

experiments with high-capacity blast furnaces and their auxiliary equipment; mechanization and automation of blast furnaces and their auxiliary equipment. It was emphasized that the Soviet Union was the first in the whole world to apply natural gas and that the combination of natural gas and oxygen blasting had improved the technical-economical conditions and reduced the coke-consumption considerably. In order to extend the use of natural gas, it was held necessary to make high-capacity oxygen equipment (units of 35,000 m³/hour capacity) at short notice. Another problem to be solved was the reconstruction of blast furnaces for operation at higher (1.5-1.8 atm) pressures. All new blast furnaces would have to be built for a volume of 2,000 m³ or more, and must operate with a pressure of 2.5 atm in the charging hole, utilizing the energy of pressureized top gas in turbine-distributors. In order to increase the efficiency of natural gas, equipment should be designed which automatically maintains the constant proportion between blast and natural gas blown through and moreover, insures a uniform distribution of blast and gas through the tuyères. Practice shows that blast furnaces with a volume of 1,719 m³ operate satisfactorily; it is, therefore, justified to increase the blast furnace volume further. Among several recommendations for improving the furnace construction

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All-Soviet Meeting of Furnacemen and Agglomeration Workers

it was suggested that the burners of the air-heating system be provided with fans having a capacity of 70,000-80,000 m³/hour, valves constructed for the air heaters, capable of operating with the blast heated up to 1,200°C, at a pressure of 4.2 atm and equipment provided for mixing, sieving, weighing and loading the charge with the aid of vibrating screens, conveyors and stationary weighers. It was found that the present state of automatization of blast furnace operation was unsatisfactory. More attention should be paid in this respect to insuring a uniform quality of the charge and a high degree of stability of furnace operation. It was pointed out that the mechanization of heavy labor in the furnace chamber did not keep pace with the increase in furnace volume. A special institute for planning the mechanization of these operations was recommended. To improve labor conditions a number of proposals was made: special equipment for changing the tuyères and slag apparatus, remote-control of charge-feeding, delivery of the charge by conveyors, hydraulic dust removal from the bunker area, pneumatic transport of dust from the collectors. For the scientific research institutes the most important tasks are: to establish the optimum methods for utilizing reducing gases, liquid fuels and oxygen, the optimum heat conditions of blast and gas pressure in the charging hole, suitable basicity of the

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S/133/60/000/012/015/015
A054/A027

All-Soviet Meeting of Furnacemen and Agglomeration Workers

slag which would make it possible to melt low-sulphur irons. Methods should be established for desulphurization of iron outside the furnace and for refining the iron in the furnace chamber. Turbine distributors should be constructed for utilizing excess pressure and the excess furnace gas; methods should be invented to increase the useful life of the furnace installation, refractory material of longer useful life should be established, equipment for grading the agglomerate and coke into fractions before loading, equipment for processing the slag at the blast furnace with mechanical collection of the semi-finished product (without ladles) should be constructed.

Card 5/5

KHODAK, L.Z. (Moskva); MALYSHEVA, T.Ya. (Moskva)

Reduction and primary slag formation processes during the forced
melting of ore and slag granules in shaft furnaces. Izv. AN SSSR.
Otd.tekh.nauk. Met.i topl. no.4:10-19 Jl-Ag '62. (MIRA 15:8)
(Blast furnaces) (Metallurgical laboratories)

KHODAK, L.Z.

Spread of combustion zones in blast furnaces. Trudy Inst. met.
no.11:16-30 '62. (MIRA 16:5)
(Blast furnaces--Combustion)

CHERNYSHEV, A.M.; GESS, B.A.; KANAVETS, P.L.; MELENT'YEV, P.N.;
KHODAK, L.Z.; SOKOLOV, G.A.; BORISOV, Yu.I.; CHERNYKH, V.I.;
Prinimali uchastiye: VAVILOV, N.S.; MAKAROV, V.G.;
KISELEV, G.P.; VOLNISTOVA, R.A.; MOREYEVA, O.R.

Testing granules made by the method of chemical catalysis
in a laboratory shaft furnace. Trudy IGI 22:70-78 '63.
(MIRA 16:11)

KHODAK, L.Z.; MALYSHEVA, T.Ya.

Changes in phase composition and the mechanism of primary
slag formation during the melting of ore-fuel granules.
Trudy IGI 22:79-92 '63. (MIRA 16:11)

MALYSHEVA, T.Ya. (Moskva); KHODAK, L.Z. (Moskva)

Reduction and primary slag formation during the smelting of
raw material prepared for blast furnace smelting. Izv. AN
SSSR, Met. i gor. delo no.6:41-47 N-D '64. (MIRA 18:3)

BORISOV, Yu.I. (Moskva); KHODAK, L.Z. (Moskva)

Certain regularities of the charge movement in blast furnaces. Izv.
AN SSSR, Met. no.3:3-10 My-Je '65. (MIRA 18:7)

BORISOV, Yu.I.; KHODAK, L.Z.

Mechanism of the flow of loose materials through an outlet. Inzh.-fiz.
zhur. 8 no.6;712-719 Je '65. (MIRA 18:7)

1. Institut metallurgii imeni Baykova, Moskva.

DORISOV, Yu.I. (Moskva); KHODAK, L.Z. (Moskva)

Mechanism of the descending of the charge into the combustion area of a blast furnace operating with a forced blow. Izv. Akad. SSSR. Met. no.6:8-13 N-D '65. (MIRA 19:1)

1. Submitted August 12, 1964.

RHODAK, N.

MADSIMUK, P.S.; RHODAK, N.

Use of blowers for feeding carbon dioxide gas to saturators. Sakh.
prom. 31 no.2:45-46 F '57. (MLRA 10:4)

1. Kupyanskiy sakharnyy zavod.
(Blowers) (Carbon dioxide) (Sugar machinery)

KHODAK, P.A.; RUVINSKAYA, I.N., Blyuss, T.S.M.

Utilizing spent alkali in the thylix process for gas purification.
Gaz.prom. 4 no.8:17-18 Ag '59. (MIRA 12:11)
(Gorlovka--Gas purification)

KHODAK, P.A.; VAS'KO, V.V.

Recovery of monoethanolamine after the purification of
coke-oven gas. Khim. prem. no. 6:511 8 '60. (MIRA 13:11)

1. Gorlovskiy azotno-tukovyy zavod.
(Ethanol) (Coke-oven gas)

KHODAK, T. I.

Fuel Abst.
Vol. 15 No. 4
Apr. 1954
Natural Solid Fuels:
Winning

✓ 2693. INKP WINCH FOR SLOPING AND STEEPLY DIPPING SEAMS. Khodak, T.I. (Ugol (Coal), Oct. 1953, 37-39). A short illustrated description is given, with diagrams, of the gear drives and electrical connections of a new winch for use with a coal cutter or cutter-loader. It is remote-controlled by the man operating the cutter. (L).

(1)

KHODAK, V.

Our present-day plans. Sil'. bud. 13 no.11:11-13 N '63.
(MIRA 17:1)

1. Predsedatel' soveta Pogrebeshchenskoy mezhkolkhoznoy
stroitel'noy organizatsii Vinnitskoy oblasti.

5.3610

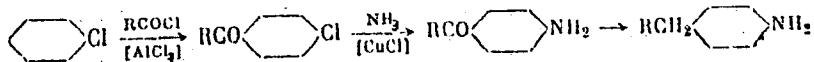
78304
SOV/79-30-3-58/69

AUTHORS: Nikolenko, L. N., Karpova, Ye. N., Khodak, V. A., Chirakadze, G. G., Borovik, V. P.

TITLE: Investigation of Aromatic Compounds With a Long Side Chain. III. Reduction of Alkyl 4-Aminophenyl Ketones According to Modified Kishner's Method

PERIODICAL: Zhurnal obshchey khimii, 1950, Vol 30, Nr 3, pp 1028-1031 (USSR)

ABSTRACT: This is a continuation of the previous work (L. N. Nikolenko, K. K. Babiyevskiy, ZhOKh, 25, 2231, 1955) on preparation of aniline homologs according to the following scheme:



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Investigation of Aromatic Compounds With
a Long Side Chain. III. Reduction of
Alkyl 4-Aminophenyl Ketones According to
Modified Kishner's Method

78304
SOV/79-30-3-58/69

A series of alkyl 4-chlorophenyl ketones (see Table 1) was obtained by the condensation of aliphatic acid chlorides with chlorobenzene in the presence of AlCl_3 . The reaction mixture was kept for 2 hr at $20-22^\circ$ and 1 additional hr at 100° . The alkyl 4-aminophenyl ketones shown in Table 2 were obtained by ammonolysis of the corresponding alkyl 4-chlorophenyl ketones. 4-Alkylanilines shown in Table 3 were obtained by reduction of the corresponding alkyl 4-aminophenyl ketones with hydrazine hydrate according to the modified Kirshner method. There are 3 tables; and 9 references, 1 U.S., 3 U.K., 2 Japanese, 3 Soviet. The 4 U.S. and U.K. references are: E. Cline, E. Reid, J. Am. Chem. Soc., 49, 3152 (1927); G. Baddeley, J. Kenner, J. Chem. Soc., 303 (1935); W. J. Hickinbottom, A. C. Waine, J. Chem. Soc., 1558 (1930); W. J. Hickinbottom, J. Hickenbottom, J. Chem. Soc., 1119 (1937).

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78304, SOV/79-30-38/69

Table 1. Alkyl 4-chlorophenyl ketones p-RCOC₆H₄Cl.

Key: (a) Yield (%); (b) mp; (c) mp of 2,4-dinitrophenyl-hydrazone.

R	a	b	c
C ₄ H ₉	80	32-32.5	175-175.3°
C ₆ H ₁₃	94	64.5-65.5	150-151
C ₈ H ₁₇	97	58-58.5	134-135
C ₁₀ H ₂₁	98	46.5-47	103.5-104.5
C ₁₁ H ₂₃	81	51.5-52	80.3-80.7
C ₁₂ H ₂₅	91	69.5-70	100-100.6

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Table 2. Alkyl 4-aminophenyl ketones $p\text{-RCOC}_6\text{H}_4\text{NH}_2$.
Key: (a) Yield (%); (b) mp.

R	a	b
C_9H_{13}	95	90-90.5
C_8H_{17}	98	91-92
$\text{C}_{10}\text{H}_{21}$	98	101.5-102
$\text{C}_{12}\text{H}_{27}$	98	104-104.5
$\text{C}_{14}\text{H}_{29}$	95	102-102.5
$\text{C}_{15}\text{H}_{31}$	99	99-100

Card 4/6

78304, SOV/79-3-58/69

Table 3. 4-Alkylanilines p-RC₆H₄NH₂. Key: (a) Yield (%) ; (b) bp (pressure in mm) and mp.

R	a	b
C ₉ H ₁₉	80	194-196 (16)
C ₁₁ H ₂₃	85	167-168 (3)
C ₁₄ H ₂₉	98	m.p. 19.5-20 m.p. 44.5-45

Card 5/6

Investigation of Aromatic Compounds With
a Long Side Chain. III. Reduction of
Alkyl 4-Aminophenyl Ketones According to
Modified Kishner's Method

78304
SOV/79-30-3-58/69

ASSOCIATION: D. I. Mendeleyev Moscow Institute of Chemical Technology
(Moskovskiy khimiko-tehnologicheskiy institut imeni
D. I. Mendeleyeva)

SUBMITTED: January 12, 1959

Card 6/6

KHODAK, V.M., inzh.

Reduction of physical and mechanical indices of slabs to
comparable values. Der. prom. 10 no. 7:14 Jl '61. (MIRA 14:7)
(Hardboard—Testing)

DRUZHININ, V.N.; FEDORISHCHEV, T.I.; KHODAK, V.M.; OSHURKOVA, I.K.

Use of hydrophobic additions obtained from turpentine industry wastes
in the manufacture of particle boards. Der.prom. 11 no.1:25-26
Ja '62. (MIRA 15:1)

(Hardboard)

KHODAK, Yu.A.

Bitumen stratification in Cambrian deposits in the Aldan District,
Yakut A.S.S.R. Dokl.AN SSSR. 105 no.3:564-565 N '55. (MLRA 9:3)

1. Institut geologicheskikh nauk Akademii nauk SSSR. Predstavлено
академиком Н.С. Шатским.
(Aldan District--Bitumen)

KHODAK, Yu A.

UESR/Cosmochemistry. Geochemistry. Hydrochemistry. D

Abs Jour : Ref Zhur - Khimiya, No. 8, 1957, 26560.

Author : Khodak, Yu.A.

Inst : Lvov University.- Inst. Geol. Sci AS USSR, Moscow

Title : Secondary Alterations of Lower Cambrian Rocks
in Aldanskiy Region of Yakut ASSR.

Orig Pub : Vopr. mineralogii osadoch. obrazovaniy. Kn.
3- 4, L'vovsk. un-t, 1956, 468 - 485.

Abstract : Some newly formed minerals were discovered
at the study of alteration of sedimentary rocks,
which have not been subject to the action of
intrusions. Quartz and chalcedony were pro-
duced in the results of concentration of the
original SiO₂ and of its transfer into the
rock layers. Argillaceous minerals served
as sources of K for the formation of felspars.

Card 1/3

KHODAK, Y. A.

Secondary minerals in the lower Cambrian deposits of the Albian region of the Volkhov A.S. S. R.

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722120007-0"

"APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722120007-0

CHODAK, Yu.A.

Alterations of Cambrian dolomitic rocks of the Aldan

intrusion. A petrographical and geochemical approach

to the problem of the age of the intrusion

and the nature of the thermal event

power distribution of altered dolomitic rocks around the

intrusion was established. Tables of rock-analysis results

substantiate the findings. Heating curves are shown for

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722120007-0"

Khodak, Yu.A.

USSR/ Geology

Card 1/1 Pub. 22 - 41/54

Authors : Khodak, Yu. A.

Title : Genesis of lower Cambrian dolomites of the Aldansk region of the Yakutsk-ASSR

Periodical : Dok. AN SSSR 106/2, 328-330, Jan 11, 1956

Abstract : Scientific data are presented on the genesis of lower Cambrian dolomites discovered in the Aldansk region of the Yakutsk ASSR. Thirteen references: 11 USSR; 1 French and 1 Eng. (1911-1955).

Institution :

Presented by: Academician N. M. Strakhov, October 23, 1955

KHODAK, Yu.A.

On fluorite from Lower Cambrian rocks of the Aldan region of the
Yakut ASSR. Dokl. AN SSSR 106 no.3:533-536 Ja '56. (MIRA 9:6)

1. Institut geologicheskikh nauk Akademii nauk SSSR. Predstavлено
академиком Н.С.Шатским.
(Aldan region--Fluorite)

KHODAK, Yu.A.; MENYAYLOV, A.A., doktor geol.-mineral. nauk, otd. red.;
SHLIMPOV, V.K., red. izd-vn; ASTAF'YEVA, G.A., tekhn. red.

[Petrographic and mineralogic characteristics of lower
Cambrian sediments in Aldan District] Petrografo-mineralo-
gicheskia kharakteristika nizhnekembriiskikh otlozhenii
Aldanskogo raiona. Moskva, Izd-vo Akad. nauk SSSR, 1960.
116 p. (MIRA 14:5)

(Aldan District--Petrology)

KHODAK, Yu.A.; SUN' SHU [Sun shu]

Main structures of northeastern China and the adjacent territory
of the Soviet Far East. Izv.AN SSSR. Ser.geol.26 no.10:97-110
0 '61. (MIRA 14:9)

1. Sovet po izucheniyu proizvoditel'nykh sil Gosudarstvennogo
nauchno-ekonomiceskogo soveta Soveta Ministrov SSSR, Moskva.
(China--Geology, Structural)
(Soviet Far East--Geology, Structural)

KHODAK, Yu.A.; CHEOBOTAREV, M.V.

Genesis of Sinaean magnesite rocks in the Lesser Khingan Mountains.
Dokl.AN SSSR 138 no.1:184-187 My-Je '61. (MIRA 14:4)

1. Sovet po izucheniyu proizvoditel'nykh sil AN SSSR i
Dal'nevostochnye geologicheskoye upravleniye. Predstavлено
akademikom D.S.Korzhinskim.
(Lesser Khingan Mountains--Magnesite)

• KHODAK, Yu.A.

Sequence and age of ancient formations in the Soviet Far East
and adjacent regions of North China. Sov.geol. 4 no.12:96-106
D '61. (MIRA 15:2)

1. Akademiya nauk SSSR.
(Soviet Far East—Geology, Stratigraphic)
(China—Geology, Stratigraphic)

ACCESSION NR: AP4009628

S/0293/63/001/003/0460/0464

AUTHOR: Khodak, Yu. A.; Kozlov, V. V.; Tomson, I. N.; Khoroshilov, L. V.

TITLE: Significance of geographic and geological methods in lunar studies

SOURCE: Kosmicheskiye issledovaniya, v. 1, no. 3, 1963, 460-464

TOPIC TAGS: lunar research, lunar geological study, lunar geographic study, lunar structure, lunar relief, lunar history, meteorite lunar theory, astronomy, moon

ABSTRACT: The report offers a brief review of lunar research to date, clarifies the significance of geographic and geological methods for future studies of lunar structure and relief, proposes close coordination of such methods (giving consideration to comparative terrestrial material) with astronomical methods, evaluates various studies of geographic and geological aspects completed thus far, and discusses the meteorite approach to an explanation of the evolution of lunar structure and relief. It is suggested that it will be impossible to clarify the origin of lunar structures and relief, or their pattern of distribution, without the participation of geologists, nor will it be feasible to compile adequate topographic, geographic or selenological-geological charts or diagrams. "The authors acknowledge the contribution of Dr. A. G. Masevich in posing the problem". Orig. art. Card 1/2

ACCESSION NR: AP4009628

has: no graphics.

ASSOCIATION: none

SUBMITTED: 09May63

DATE ACQ: 30Jan64

ENCL: 00

SUB CODE: AS

NO REF Sov: 019

OTHER: 039

Card 2/2

ACCESSION NR: AP4009629

8/0293/63/001/003/0465/0471.

AUTHOR: Khodak, Yu. A.

TITLE: The most important structural elements of the Moon

SOURCE: Kosmicheskiye issledovaniya, v. 1, no. 3, 1963, 465-471

TOPIC TAGS: moon, moon structural elements, lunar topography, albedo, lunar albedo, lunar relief, moon map

ABSTRACT: The character and location of lunar topographic features are analyzed to trace the principal structural elements of the Moon. Recent photographs, maps, and other data on lunar relief are employed. The paper focuses on three fundamental structural elements said by the author to have been formed about 300 million years ago: 1) a meridionally extended ancient massif embracing the western portion of the far side and the southern portion of the visible side; 2) the Great Belt of large depressions (seas) located within the massif and extending convexly in the form of a semi-circle to the north pole from the South Sea to the Sea of Moisture; 3) the meridional belt of large depressions of the far side of the Moon, located on the edge of an uncharted lunar area. Within these, it is possible to

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ACCESSION NR: AP4009529

trace the evolution of blocks limited by a system of global depth fractures extending in four directions for thousands of kilometers: meridional, submeridional, latitudinal, and sublatitudinal. The geophysical significance of these fractures (breaks) is discussed, and the most ancient systems are traced in detail. The author states that approximately 100-150 million years ago the Moon had to a large degree inherited its ancient structural plan; the central meridional zone of fractures of the southern hemisphere was the most mobile, while the far side was fairly stable. In succeeding periods a further differentiation of structural elements along the depth fault zones created the present enormous blocks. A demonstration of this hypothesis is given. The author cites the evidence of A. V. Peyve ("Struktura zemnoy kory i deformatsii gornykh porod," Izd-vo AN SSSR, 1969) and G. N. Katterfel'd (Izv. Vses. geogr. o-va, v. 91, no. 272, 1952) on the existence of a block structure on the Earth and Mars as reason for investigating the possibility of a general law operative in the development of a hard core in these bodies. "The author wishes to express his gratitude to Doctor of Physics and Mathematics A. G. Masevich for her help in conducting the studies." Original article has: 2 figures.

Card 2/bz

KHODAI, Yu.A.

Principal structural elements of the moon and the geographical
and geological methods of studying them. Izv. Kom. po fiz. plan.
no. 4:10-23 Ag '63. (MIRA 18:5)

J. Laboratoriya osadochnykh poleznykh iskopayemykh Gosudarstvennogo
geologicheskogo komiteta SSSR.

A. KHODAK, Yu.A.

Geological of the Lesser Khingan Mountains and its position in the
structure of the Far East. Sov. geol. 6 no.6:20-31 Je '63.
(MIRA 1.6:7)

1. Laboratoriya osadochnykh poleznykh iskopayemykh AN SSSR,
(Khingan Mountains—Geology, Structural)

KHODAK, Yu.A.

Main structural elements of the moon and the importance of geographical and geological methods in lunar explorations. Izv. AN SSSR. Ser. geol. 28 no.8:11-22 Ag '63. (MIRA 17:2)

KHODAK, Yu.A.; CHEBOTAREV, M.V.

Ancient formations in the Amur Valley. Sov. geol. 7 no.1:
79-94 Ja '64. (MIRA 17:6)

1. Dal'nevostochnoye geologicheskoye upravleniye i Laboratoriya
osadochnykh poleznykh iskopayemykh Gosudarstvennogo geologi-
cheskogo komiteta SSSR.

KHODAK, Yu.A.

Structural and genetic characteristics of the iron-manganese
mineralization of the Karashal type (central Kazakhstan). Trudy
SNIIGGIMS no.35:218-227 '64. (MIRA 18:5)

YEROSHICHENKOV, V.A.; KHODAK, Yu.A.; GRIBOV, Ye.M.; SYNGAYEVSKIY, Ye.D.

Association of clay minerals in the Upper Famennian rocks and ores
of the Dzhail'ma trough. Dokl. AN SSSR 164 no.4:906-909 O '65.
(MIRA 18:10)

I. Laboratoriya osadochnykh poleznykh iskopayemykh AN SSSR. Sub-
mitted May 12, 1965.

ACC NR: AR6035075

SOURCE CODE: UR/0169/66/000/008/G001/C001

AUTHOR: Khodak, Yu. A.

TITLE: Defining the main features of the structure and development of the Moon
and their significance in the elucidation of principles of geological phenomena

SOURCE: Ref. zh. Geofizika, Abs. 8G1

REF SOURCE: Sb. Materialy k Soveshchaniyu Obshchiye zakonomern. geol.
yavleniy, 1966, Vyp. 1. L., 1965, 197-202

TOPIC TAGS: moon, geology, selenography, earth, planet, map

ABSTRACT: In order to find the basic magnetic and block structure of the Earth, it is necessary to find the principles of the distribution of lunar ring structures and their connection with large and local blocks. The basic problem in the study of the structure of the Moon is the development of the stratigraphic scale of its formations. A structural-selenological map of the visible part of the Moon at a scale of 1:2,500,000 was compiled. Four structural strata are seen on the map. The areas of an emerging, mechanically eroded, ancient basement were noted. The

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UDC: 550.311

ACC NR: AR6035075

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study of the structure of the planetary body of the moon will help to determine the general planetary structure of the Earth as a planet. [Translation of abstract]

SUB CODE: 03, 08/

Card 2/2

KHODAKIN, N. I.

Soshnikova, M. N. and N. I. Khodakin "The Aetiological Agent of a Peculiar Form of Encephalitis," Dok. AN, 47, No. 5, 1945. Mbr., Uzbek Inst. Microbiology & Epidemiology, Tashkent, -1944-049.

KHODAKOV, A.

A transformerless rectifier. Radio no, 2135 P '63.

(MIRA 16:2)

(Electric current rectifiers)

S/196/63/000/001/006/035
E193/E383

AUTHORS: Protsenko, P.I., Khodakov, A.A., Mirskaya, Ye.Z. and Venerovskaya, L.N.

TITLE: Physicochemical parameters of nitrites and nitrates of alkali and alkaline-earth metals with ferroelectric properties

PERIODICAL: Referativnyy zhurnal, Elektrotehnika i energetika, no. 1, 1963, 17, abstract 1 'B55. (In collection: Segnetoelektriки (Ferroelectrics), Rostov-na-Donu, Rostovsk. un-t, 1961, 21-26)

TEXT: In connection with the possible application of ferroelectrics as nonlinear elements in conjunction with electrolumino-phors, it is desirable to have available ferroelectrics characterized by low ϵ , this property being necessary to ensure their compatibility with electroluminophors. With this in view, a study was conducted of crystals of those nitrites and nitrates of alkali and alkaline-earth metals that possess ferroelectric properties; the experimental specimens were crystallized out of aqueous solutions or grown by the Bridgman method from their melts. Thermal Card 1/5

S/196/62/000/001/006/035
Physicochemical parameters ; E193/E383

analysis of a large number of nitrates and nitrites enabled the authors to obtain more accurate data on their melting points, to establish the existence of polymorphic transformations and to determine the transformation temperatures (these data being reported in the form of a table). It was shown that single crystals of sodium nitrite (NaNO_2) in the direction of the G axis constituted ferroelectrics with $\Theta \sim 457^\circ\text{K}$, i.e. 164°C (see Fig. 1), the magnitude of ϵ at Θ being more than 100 times higher than that at room temperature. The magnitude of spontaneous polarization, determined by pyroelectrical measurements, was found to be about $7 \mu\text{C}/\text{cm}^2$. Typical hysteresis loops were observed at 413°K (140°C) at 50 c.p.s. High values of coercive fields at room temperature were established. A study of the dependence of ϵ of NaNO_2 on temperature and the intensity of the DC field E showed that ϵ decreased with increasing E at temperatures lower than Θ , being independent of E at Θ . Dilatometric measurements showed that the temperature coefficient of linear expansion α of NaNO_2 was of the order of $10^{-4} - 4 \times 10^{-5} \text{ deg}^{-1}$, and that the temperature-dependence of α differed from that typical for ferroelectrics. A

Card 2/5

S/196/62/000/001/006/035
E193/E383

Physicochemical parameters

domain structure was observed which disappeared at temperatures higher than θ and was not restored on cooling below θ . Single crystals of sodium, rubidium, caesium and thallium nitrates had phase-transformations in the temperature range between room temperature and the melting point. The transformation of sodium nitrate from the second phase (with an orthorhombic structure of aragonite) to the first phase (with the calcite structure) took place on heating above 403 $^{\circ}$ K (130 $^{\circ}$ C); on cooling below 397 $^{\circ}$ K (124 $^{\circ}$ C) the first phase changed into ferroelectric third phase, which remained stable down to 383 $^{\circ}$ K (110 $^{\circ}$ C) and then changed to the second phase. The transformation of sodium nitrate to its ferroelectric phase was accompanied by a decrease in ϵ . Transformation from hexagonal to cubic modification took place at 434 $^{\circ}$ K (161 $^{\circ}$ C) in rubidium nitrate; a change from cubic to rhombic modification taking place at 492 $^{\circ}$ K (219 $^{\circ}$ C); a phase-transformation in this compound was observed also at 564 $^{\circ}$ K (291 $^{\circ}$ C). Rubidium nitrate had no ferroelectric properties in the temperature interval studied. A phase-transformation took place in caesium nitrate at 427 $^{\circ}$ K (154 $^{\circ}$ C) but no maxima were observed near the transformation temperature. Two phase-transformations were observed in thallium nitrate (see Card 3/5)

S/196/62/000/001/006/035
E195/E383

Physicochemical parameters

Fig. 2): $\gamma \rightarrow \beta$ transformation at 348°K (75°C) and $\beta \rightarrow \alpha$ transformation at 418°K (145°C). The increase in ϵ observed on heating thallium nitrate was attributed to the increase in conductivity. No ferroelectric properties were observed in barium nitrate, a $(\text{Tl}-\text{Ba})\text{NO}_2$ complex and certain other nitrates. There are 5 figures and 3 references.

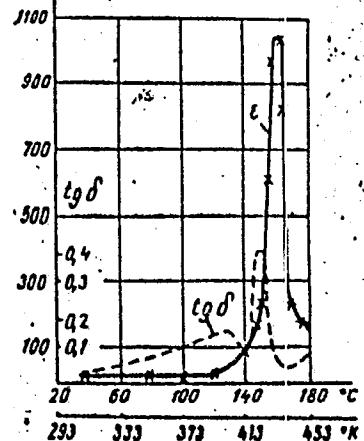
[Abstracter's note: Complete translation.]

CAPTION to Fig.1:

Temperature-dependence of ϵ and $\tan \delta$ of NaNO_2 at $f = 1 \text{ Mc/s}$

Card 4/5

Fig. 1:



S/019/61/000/024/031/088
A156/A126

AUTHORS: Sholokhovich, M.L.; Khodakov, A.L.

TITLE: Seignettelectric monocrystals

PERIODICAL: Byulleten' izobreteniy, no. 24, 1961, 30 .

TEXT: Class 21g, 1102. No. 143477 (688328/26 of December 8, 1960).
Seignettelectric monocrystals based on barium titanate, the distinctive feature
of which consists in that for the purpose of increasing the nonlinearity of
their characteristics and raising the rectangularity of their hysteresis loop,
1 - 2% barium hafnate is added to the barium titanate.

Card 1/1

L 15629-55 EMT(1)/EP4(-)-Z/EMT(m)/FSC(t)/FPC(1)/C/1812-0000
ASD
ACCESSION NR 7 10478741 3/19 S/000722120007-0

SOURCE: Alm, Minimya, Abs. 12B466

AUTHOR: Kranarov, P. F.; Khodakov, A. L.; Sholobovitch, V.
Fesenko, Ye. G.

TITLE: Monocrystals of solid solutions of strontium and lead titanates

CITED SOURCE: Sb. Segnetoelektriki. Rostov-na-Donu, Rostovsk, un-t,
1961, 5-11

TOPIC WORDS: solid solution, strontium, lead, strontium
lead titanate, monocristalline structure

TRANSLATION: The fusion diagram for the system K_2TiO_3 - $PbTiO_3$ has been studied and the formation of a continuous solid solution $(K_2TiO_3)_x(PbTiO_3)_{1-x}$ has been established. The effect of the position of the metal atom in the crystal lattice on the electrical conductivity, especially constructed galvanometer with the permitted range of variation in samples of 0.1-100 ohms.

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L 15629-65
ACCESSION NR: AR3010278

transition temperature of 512° for monocrystals of PbTiO_3 was determined by the same method. The Curie point for monocrystals of solid solutions is close to the data known for polycrystalline samples. The refractive index for monocrystals ($\text{Pb--Sr}_x\text{TiO}_3$) changes in a nonmonotonic fashion within the limits of 2.35 (for SrTiO_3) to 2.70 (for PbTiO_3).

SUB CODE: MM, SS

ENCL: 00

Card 2/2

ACCESSION NR: AR4042161

S/0196/64/000/005/B019/B019

SOURCE: Ref. zh. Elektrotehnika i energetika, Abs. 5B83

AUTHOR: Lezgintseva, T. N.; Khodakov, A. L.

TITLE: Influence of slight impurities of iron on the dielectric properties of solid solutions of barium titanate and stannate

CITED SOURCE: Izv. Leningr. elektrotekhn. in-ta, vy*p. 51, 1963, 260-267

TOPIC TAGS: barium titanate, barium stannate, dielectric property solid solution

TRANSLATION: The dependence of ϵ on the intensity of a variable electric field E (up to 10 kv/cm), reversible ϵ (E varied up to 8 kv/cm) was studied at 300 kc, hysteresis loop and dependence on temperature of ϵ and $\tan \delta$ at 300 kc from 20 to 140°C for ceramic solid solutions of $BaTiO_3 - BaSnO_3$ with 0; 3; 6; 9 and 12 mole % $BaSnO_3$ and 0; 0.1; 0.2; 0.4; 0.7; 1 mole % Fe_2O_3 . Introduction of additions of Fe_2O_3 leads to a sharp lowering of the nonlinear properties of solid solutions; this is,

-Card 1/2

ACCESSION NR: AR4042161

especially noticeably for compositions containing 6 mole % BaSnO₃. In solid solutions with additions of iron, θ shifts in the direction of low temperatures, the more noticeably, the higher the concentration of Fe, while ϵ is also lowered at θ . These effects are more noticeable in solid solutions baked directly from a mixture of BaTiO₃, BaSnO₃, and Fe₂O₃. The influence of Fe impurities on θ and ϵ in pure samples of BaTiO₃ is noticeably less than in solid solutions alloyed with the same concentration of Fe. For the manufacture of ferroelectric-ceramics and single crystals with the sharpest expressed nonlinear properties, it is proposed, to avoid materials containing Fe. Three illustrations. Bibliography: 4 references. [Rostov-on-Don State University]

SUB CODE: IC, EM

ENCL: 00

Card

2/2